Improving the optical efficiency of white light-emitting diodes based on phosphor-in-glass by a dual-layer remote phosphorus structure with the application of LiLaO$_2$:Eu$^{3+}$ and CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$

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ABSTRACT
While the remote phosphor structure is not an appropriate solution for white light emitting diode (WLED) color uniformity, it is more advantageous for the luminous output of WLED than the conformal phosphor or in-cup phosphor structures. Acknowledging the ability of the remote phosphor structure, many studies have been carried out to surmount the color quality disadvantage of this structure. A dual-layer remote phosphor configuration is proposed in this research paper to acquire better color quality for WLEDs through heightening the color rendering index (CRI) and the color quality scale (CQS). The color temperature of the WLED packages this study is 8500 K. By inserting a layer of green CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$ or red LiLaO$_2$:Eu$^{3+}$ phosphor on the yellow YAG:Ce$^{3+}$ phosphor layer, the phosphor structure configuration can be constructed. Then, to get the best color quality, the concentration of added phosphor LiLaO$_2$:Eu$^{3+}$ would be changed. The findings showed the rise of CRI and CQS along with the LiLaO$_2$:Eu$^{3+}$, which implies the influence of LiLaO$_2$:Eu$^{3+}$ to the growth of red light components within WLEDs packages. The greater the concentration of LiLaO$_2$:Eu$^{3+}$ is, the more the CRI and CQS increase. Meanwhile, the luminous flux gains from the green phosphor CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$.

Keywords: CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$
Color quality
LiLaO$_2$:Eu$^{3+}$
Lumen output
Mie-scattering theory
WLEDs

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1. INTRODUCTION
As the lighting industry has higher requirements for lighting solution these days, phosphor converted white light emitting diodes (pc-WLEDs) are considered to be the fourth viable light source generation and can be an alternative to the conventional one [1]. In our daily life, the popularity of white light-emitting diodes is undeniable. In several ways, such as architecture, street lighting, backlogging, and more, they are used. However, there are two principal factors that white LEDs require enhancements to widespread their usage: the efficiency in light extraction and the angular uniformity of associated color temperatures [2]. Further breakthroughs in lumen output and color quality of WLEDs are therefore necessary for meeting high market requirements as well as life application requirements [3]. The combination of blue light from the reverse red phosphorus with yellow light from the LED chip is one of the most common methods for enhancing this potential light source. It is undeniable that the factors considered to be the determining effects

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on the efficiency of lumen output, especially the color rendering index (CRI) are the LEDs’ structure and the arrangement of phosphor layers [4]-[8]. Therefore, there are several phosphor coating methods commonly utilized for LEDs’ development, including dispensing and conformal coating approaches [9], [10]. However, the color quality of these structures is low due to the degraded light conversion of the phosphors. The cause of this degradation is the rise in temperature at the interface between the LED chip and phosphor layer, where the yellow emitting phosphor particle and the chip directly contact each other. Therefore, a decrease in the thermal outcome is essential to improve the efficiency of phosphor layers and prevent the irreversible destruction of phosphors. In several previous researches, it is showed that the remote phosphor structure can reduce the heat effect as the phosphor layers of this structure are placed distantly from the LED chip which is also known as a heat source. There is an appropriate gap between the layers of phosphors and the LED chip that could restrict the backscattering and light circulation within the LEDs. Therefore, this is an ideal solution for controlling the heat from the LED chip in order to substantially increase the lumen efficacy and color quality of the LEDs [11]-[16]. The remote phosphor structure can manage to meet the requirements of regular lighting but for those of other illumination applications, it cannot completely fulfill. Thus, the development of the next WLED’s generation is understandable and essential. For the next upgrade LEDs packages, several new remote phosphor configurations are introduced to reduce the backscattering events and improve the luminous efficiency. In other words, the focus of using these remote phosphor models is to limit the events in which the phosphors scatter back to the chip. A research showed that the light path from the chip to the surface of a LED package can be redirected by a structure of an inverted cone lens encapsulant combining with a surrounding ring remote phosphor layer, which leads to the reduction in the light loss caused by internal reflection [17]. In addition, if a patterned remote phosphorus configuration has a clear region, which is not phosphorus-coated on the surrounding surface, in the perimeter area, high uniformity of angular-dependent associated color temperature and stability of color could be achieved [18]. Moreover, it is also useful to use patterned sapphire substrate in remote phosphorus as this pattern could increase the associated color temperature uniformity in a far-field pattern, compared to that in the original pattern [19]-[21]. The dual-layer remote phosphor structure is then proposed to increase the light performance of LED packages. Previous studies have focused on improving the color uniformity and luminous output of WLEDs when using the remote phosphor structure, but have focused only on low color temperature single-chip WLEDs. Meanwhile, it is a difficult and complicated task to improve the lighting parameters for WLEDs with a high color temperature. Thus, no previous study has shown any comparison between the efficacies of different dual-layer phosphor configurations. Thus, to optimize the color quality and luminous flux of WLEDs, it is difficult for manufacturers to determine an optimal remote phosphor layer structure.

This research paper introduces two distinct dual-layer remote phosphor structures to achieve the objective of optimizing the color consistency of WLEDs at a high color temperature of 8500 K. The first concept of this study is the use of the green-emitting phosphor CaSO₄:Ce³⁺, Mn²⁺ layer to increase the green-light portion in WLEDs, then heightening the luminous flux. In order to increase the CRI and CQS values, the second one uses the red LiLaO₂:Eu³⁺ phosphor layer to achieve a growth in the red light component. In addition, this research includes a detailed explanation of LiLaO₂:Eu³⁺ chemical’s composition, which has an important impact on the lighting performances of WLEDs. Besides, the results of this research have shown that when the phosphor LiLaO₂:Eu³⁺ is added, there is a noticeable improvement in the values of CRI and CQS. However, when the concentrations of blue or red phosphorus rise excessively, the concentration of CaSO₄:Ce³⁺, Mn²⁺ and LiLaO₂:Eu³⁺ must be modified to an appropriate amount to avoid a dramatic reduction in the quality of color or lumen output. There are three major variations in having a green or red phosphor film along with the existing YAG: Ce³⁺ yellow phosphor layer from the results of the studies. First, the increase of components of blue or red light will enhance the spectrum of white light, the most important point in enhancing the quality of the color. Second, depending on the concentrations of the additional phosphor layers, the scattering and the transmission of lights in WLEDs are opposite. That is why it is an important job to increase the luminous flux of WLEDs to establish an appropriate amount for phosphor layers’ concentrations.

2. DETAIL OF EXPERIMENT AND SIMULATION

2.1. Preparation of phosphor materials

Before constructing the remote phosphor configurations, red and green phosphors, LiLaO₂:Eu³⁺, and CaSO₄:Ce³⁺, Mn²⁺, must be prepared with the ingredients having mole percent and weight listed in Table 1 and Table 2. To generate LiLaO₂:Eu³⁺, it is important to strictly follow a 6-step process consisting of mixing ingredients, dehydrating, firing, powderizing, re-firing, and re-powderizing. First, slurry all the ingredients in methanol to mix them together. After the combination become homogeneous, let it dry in air, and then grind the dry mixture into powder. Next, the powder is fired in open alumina crucibles at 600 °C with air flows. The powder now is solidified and need to be powdered before being re-fired in the same machine.
For the second firing, the temperature is set at 1000 °C, and the time is 1 hour. Once the second firing process finishes, grinding the product to achieve LiLaO₂:Eu³⁺ phosphor. The characteristics of LiLaO₂:Eu³⁺ phosphor can be listed is being as: Emitting red emission at the emission peak from 1.775 to 2.02 eV. It is a must to put the final product in a well-closed container for a good preservation and avoiding contamination. The chemical composition of LiLaO₂:Eu³⁺ green phosphor is detailed in Table 1. The preparation of CaSO₄:Ce³⁺, Mn²⁺ consists of four firing processes in capped quartz tubes with N₂, but first, the ingredients must be mixed by drying grinding or milling. After this step, the firing stages will be started. The first firing of the material will be carried out for an hour at 1000°C. When the firing time is over, the product will be powdered. Then, it will be fired for the second time for an hour at 800 °C. When this process is finished, the product will be taken out and powdered once again. Next, it is fired for the third and fourth time at the same temperature 700°C. The outcomes are also need to be powdered. The final product should be washed in water several times and dried for the last time. The obtained CaSO₄:Ce³⁺, Mn²⁺ will have emission color of yellow-green and emission peak at 2.35 eV.

### Table 1. Ingredients of LiLaO₂:Eu³⁺ phosphor composition

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Mole %</th>
<th>By weight (g)</th>
</tr>
</thead>
<tbody>
<tr>
<td>La₂O₃</td>
<td>95 (of La)</td>
<td>155</td>
</tr>
<tr>
<td>Eu₂O₃</td>
<td>5 (of Eu)</td>
<td>8.8</td>
</tr>
<tr>
<td>Li₂CO₃</td>
<td>101 (of La)</td>
<td>37.4</td>
</tr>
</tbody>
</table>

### Table 2. Ingredients of CaSO₄:Ce³⁺, Mn²⁺ phosphor composition

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Mole %</th>
<th>By weight (g)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CaSO₄</td>
<td>88.5</td>
<td>121</td>
</tr>
<tr>
<td>CeO₂</td>
<td>2.5</td>
<td>4.3</td>
</tr>
<tr>
<td>MnCO₃</td>
<td>6</td>
<td>6.9</td>
</tr>
<tr>
<td>(NH₄)₂SO₄</td>
<td>~10</td>
<td>13.2</td>
</tr>
</tbody>
</table>

#### 2.2. Simulation process

Two different dual-layer phosphor configurations, including the green-yellow configuration (GYC) and the red-yellow configuration (RYC), were presented in this article. Both the GYC and RYC structures are built on blue chips with two phosphor layers. In particular, the phosphorus layer CaSO₄:Ce³⁺, Mn²⁺ is above the YAG:Ce³⁺ yellow phosphor layer in the GYC structure, as shown in Figure 1 (a). In the meantime, the phosphorous layer of LiLaO₂:Eu³⁺ is set above YAG:Ce³⁺ yellow phosphorous layer in the RYC structure, as shown in Figure 1 (b). In Figure 1 (c), the measured spectra of the red-emitting LiLaO₂:Eu³⁺ phosphor is displayed. The purpose of applying the GYC and RYC is to acquire the enhancement in the quality of color and lumen efficacy of WLEDs. This goal can be achieved by pushing the component of green scattering and red light inside the packages of the WLED. Nevertheless, the concentrations of CaSO₄:Ce³⁺, Mn²⁺, and LiLaO₂:Eu³⁺ phosphors must be adjusted accordingly to succeed in doing this. The concentrations of green CaSO₄:Ce³⁺, Mn²⁺, and red LiLaO₂:Eu³⁺ phosphors shift in the opposite direction to that of yellow YAG:Ce³⁺ phosphors, as can be seen in Figure 2. The average CCTs will be maintained by this distinct change which will affect the scattering and absorption of phosphorus layers in WLEDs. This would undoubtedly have a great influence on the consistency of the color and the luminescence of the LED lamps. The choice of the concentration of these phosphorous CaSO₄:Ce³⁺, Mn²⁺, and LiLaO₂:Eu³⁺ is therefore a deciding factor for improving the color quality of the WLED. Specifically, when the concentrations of CaSO₄:Ce³⁺, Mn²⁺, and LiLaO₂:Eu³⁺ increase from 2% to 20% wt., YAG:Ce³⁺ concentration decreases in order to maintain the average CCT, even the high color temperature of the WLEDs which is 8500 K.

![Figure 1. Illustrations of, (a) GYC, (b) RYC, (c) the measured spectra of the red-emitting LiLaO₂:Eu³⁺ phosphor](image-url)
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Figure 2. The change of phosphor concentration of RYC (above) and GYC (below) for keeping the average CCT, (a) red LiLaO$_2$:Eu$^{3+}$, (b) green CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$.

On top of that, as displayed in Figure 3, the effect on the WLED spectrum caused by concentration of red LiLaO$_2$:Eu$^{3+}$ is noticeable. The red-yellow configuration demonstrates a rise in spectral emissions in three distinct regions compared to the green phosphorus CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$ structure. In Figure 3(a), as the CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$ concentration increases, the intensity increases in the spectral regions of 420 nm-480 nm and 500 nm-640 nm. In other words, the rise in the emission spectrum indicates a greater luminous flux in these two regions. In addition, there are further internal blue-light scattering events for CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$, which implies that the phosphor scattering within the LEDs is improved, and the copper color is benefitted. Meanwhile, with the rise in the concentration of LiLaO$_2$:Eu$^{3+}$, the emission spectra show an improvement in 648 nm-738 nm region. However, without the increase emission spectra in the two similar regions of 420 nm-480 nm and 500 nm-640 nm, this development is unimportant, because the increase intensity of these two spectral regions will be beneficial to the lumen output of blue light (blue-light scattering). The growth in emission spectra along with the increase of color temperature is another result that can be seen from Figure 3(b). In other words, the higher the temperature of the color, the higher the spectral emission becomes. Hence, it is possible to achieve better the color and optical quality. This result is an essential reference for LEDs manufacture with the implementation of LiLaO$_2$:Eu$^{3+}$ phosphor, especially when it is difficult to control the quality of WLEDs with high temperature. In brief, the research defines that red phosphor LiLaO$_2$:Eu$^{3+}$ will improve the color quality of WLEDs having high color temperature (8500 K). Based on the requirements of manufacturers, the selection of the structure can be made. If they want to manufacture the WLEDs with high color quality, a slight decrease in luminous flux is acceptable.

Figure 3. Emission spectra of RYC (above) and GYC (below), (a) red LiLaO$_2$:Eu$^{3+}$, (b) green CaSO$_4$:Ce$^{3+}$, Mn$^{2+}$.
3. COMPUTATION AND DISCUSSION

Compared to the real one, the color rendering index (CRI) is a measurement used to examine how the light source reveals the true color of the objects. The color imbalance occurs depending on the color principle of the three main colors green, red and yellow when the green light component is higher than the others. As a result, the WLED’s light chromatic performance is negatively affected and then, reducing their color integrity. In Figure 4, when the green phosphor CaSO₄:Ce³⁺, Mn²⁺ is added, the charts illustrate the decline of CRI. Specifically, CRI decreases gradually when the concentration of this phosphorus increases from 2% to 20%. However, CRI is not an ideal index when being compared to color quality scale (CQS), since it does not evaluate all dimensions of the quality of illumination. These reductions in CRI are therefore acceptable when using CaSO₄:Ce³⁺, Mn²⁺. It is a mixture of the three fundamental components, including CRI, individual desires, and color coordinates, in terms of CQS. Thus, CQS is a more significant objective and difficult to accomplish. According to the results from Figure 5, CQS maintains its high values while the CaSO₄:Ce³⁺, Mn²⁺ concentration does not exceed 8 %. In comparison, the CQS declines dramatically as the CaSO₄:Ce³⁺, Mn²⁺ concentration rises from 8 % to 20 %. Therefore, an appropriately chosen concentration of LiLaO₂:Eu³⁺ will fluctuate from 2 % to 8 % if the target is to get better color quality, after considering emitted luminous flux.

In the meantime, CRI demonstrates a light rise with the presence of red phosphor LiLaO₂:Eu³⁺. In other words, with the increase of LiLaO₂:Eu³⁺ concentration, CRI is increased, as seen in Figure 4 (above). In

![Figure 4. The color rendering index as a function of the concentration of, (a) red LiLaO₂:Eu³⁺, (b) green CaSO₄:Ce³⁺, Mn²⁺](image)

![Figure 5. The color quality scale as a function of the concentration of, (a) red LiLaO₂:Eu³⁺, (b) green CaSO₄:Ce³⁺, Mn²⁺](image)
the remote phosphor structure, the force behind this upward trend of CRI when using red phosphor LiLaO$_2$:Eu$^{3+}$ is its absorption feature. Not only can the blue lights from LED chips be absorbed by this red phosphor, but also the yellow light. In addition, LiLaO$_2$:Eu$^{3+}$ can transform the blue light into the red light while absorbing blue light. Therefore, the absorption of blue light is better than the yellow one because of its absorption feature, while this red phosphor absorbs two types of light. Consequently, the addition of LiLaO$_2$:Eu$^{3+}$ has increased the red light component, leading to a higher index of color rendering (CRI). For an effective modern WLED lamp, CRI is a key parameter, because if an LED package has a high color rendering index, its price would be higher than the others. However, by using LiLaO$_2$:Eu$^{3+}$, the cost of production can be reduced, contributing to the popularity of its use in LED manufacturing. Nevertheless, researchers have concentrated on CQS, a more synthetic index to measure the color quality of LEDs than CRI, as mentioned above. Figure 5 shows that, similar to the CRI, red phosphor LiLaO$_2$:Eu$^{3+}$ will boost CQS values. The first chart in Figure 5 shows that the dramatic rise in CQS is due to the increase in concentration of LiLaO$_2$:Eu$^{3+}$. It can therefore be concluded that adding the phosphor layer LiLaO$_2$:Eu$^{3+}$ can increase the quality of white light color for WLEDs built with a dual-layer phosphorus structure. This is one of the most valuable findings for manufacturers to consider when it comes to the objective of enhancing the color quality. However, it should not be ignored that LiLaO$_2$:Eu$^{3+}$ is somehow a downside to the lumen output of WLEDs. Next, the scientific model in which the mathematic framework of transmitted blue lights and converted yellow lights will be presented to investigate their activities in the double-layer phosphor structure. In addition, a major development in the efficacy of LED lights can be attained from this model. For a single layer remote phosphorus package with a phosphorus layer thickness of $2h$, the transmitted blue light and converted yellow light are expressed as the two expressions below:

$$PB_1 = PB_0 \times e^{-2\alpha_1 h}$$  \hspace{1cm} (1)

$$PY_1 = \frac{1}{2} \frac{\beta_1 \times PB_0}{\alpha_{y1} - \alpha_{y1}} (e^{-2\alpha_{y1} h} - e^{-2\alpha_{y1} h})$$  \hspace{1cm} (2)

For the dual-layer remote phosphorus structure with the phosphor layer thickness of $h$, the transmitted blue light and converted yellow light are defined as:

$$PB_2 = PB_0 \times e^{-2\alpha_2 h}$$  \hspace{1cm} (3)

$$PY_2 = \frac{1}{2} \frac{\beta_2 \times PB_0}{\alpha_{y2} - \alpha_{y2}} (e^{-2\alpha_{y2} h} - e^{-2\alpha_{y2} h})$$  \hspace{1cm} (4)

$h$ is the thickness of each coating of phosphorus. Subscripts "1" and "2" express remote phosphor structures with single and dual layers. $\beta$ indicates the conversion coefficient for the conversion of blue light into yellow light, while $\gamma$ presents the yellow light reflectance coefficient. $PB$ is the blue light intensity, and $PY$ is the yellow light intensity. The combination of $PB$ and $PY$ demonstrated the light intensity of the blue LED, presented by $PB_0$. In addition, $\alpha_2$ and $\alpha_2$ separately describe the fractions of the blue and yellow lights’ energy losses, during the light multiplication in the phosphor film, respectively. The illumination efficacy of WLEDs is higher with the double-layer phosphorus structure than with the one-layer structure:

$$\frac{(PB_2 + PY_2) - (PB_1 + PY_1)}{PB_1 + PY_1} > 0$$  \hspace{1cm} (5)

The scattering of the phosphorus particles was defined based on the Mie theory [22], [23]. Moreover, the scattering cross section of $C_{ext}$ for spherical particles is estimated with the support from the Mie theory. Furthermore, applying the Lambert-Beer law [24], [25] can help to compute the transmitted light power:

$$I = I_0 \exp(-\mu_{ext} L)$$  \hspace{1cm} (6)

$I_0$ is the incident light intensity in this expression, $L$ indicates the phosphor layer’s thickness (mm), and $\mu_{ext}$ is the coefficient of extinction. In addition, $\mu_{ext}$ can be calculated by: $\mu_{ext} = N \cdot C_{ext}$, in which $N$ ($\text{mm}^{-1}$) is defined as the distribution of particle number density, and $C_{ext}$ ($\text{mm}^{-1}$) presents the cross-section of

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phosphorus particles for extinction. Expression (5) demonstrated that the luminous efficiency of remote phosphor pc-LEDs with dual layers is higher than single layer of phosphorus. Thus, the lumen efficacy of the dual-layer remote phosphor layer added in LED packages has been carried out and proved by this study.

In Figure 6, as CaSO₄:Ce³⁺, Mn²⁺ is raised from 2% wt. up to 20% wt., we can observe the development of luminous flux. In comparison, the luminous flux of the dual-layer remote phosphor structure shows a sharp decline with the rise in LiLaO₂:Eu³⁺ phosphor concentration. The reduction factor $\mu_{ext}$ is obviously in direct proportion to the LiLaO₂:Eu³⁺ concentration, according to the Lambert-Beer law, though in inverse proportion to the energy of light transmission. Therefore, if the thicknesses of both phosphor films are fixed, as the concentration of LiLaO₂:Eu³⁺ increases, the emitted photoluminescence could decrease. However, the dual-layer phosphor structure with the additional red LiLaO₂:Eu³⁺ phosphor yields better luminous flux than the single-layer phosphor structure does, as demonstrated through the calculations above. What's more, the CRI and CQS values also gain from this red phosphor. Therefore, by using LiLaO₂:Eu³⁺, these benefits are obtained and this reduction in lumen output is entirely acceptable. Based on the objective they wish to achieve manufacturers will select an appropriate concentration of LiLaO₂:Eu³⁺ to apply for WLEDs mass production.

![Figure 6](image)

Figure 6. The lumen output as a function of the concentration of, (a) red LiLaO₂:Eu³⁺, (b) green CaSO₄:Ce³⁺, Mn²⁺.

4. CONCLUSION

The effects of the green CaSO₄:Ce³⁺, Mn²⁺ and red LiLaO₂:Eu³⁺ phosphors in the dual-layer phosphor configurations on the optical features of WLEDs, including CRI, CQS and luminous flux, are shown in this study. By applying the Mie-scattering theory and the law of Lambert-Beer, it can be inferred from the study that LiLaO₂:Eu³⁺ is a suitable phosphor material for improving color quality, whereas CaSO₄:Ce³⁺, Mn²⁺ is the perfect choice for WLEDs to increase luminous flux. This result is not only true for low color temperature WLEDs, but also for those with a high 8500 K color temperature. This study has also accomplished the objective of enhancing the white-light chromatic quality that the remote phosphor structure has been struggling to acquire. A slight drawback in terms of luminous flux, however, still exists: when CaSO₄:Ce³⁺, Mn²⁺ or LiLaO₂:Eu³⁺ concentrations increase excessively, the quality of color or luminous flux tends to decrease dramatically. Therefore, it is very important to determine the required phosphorus concentration based on what the manufacturers aim to achieve. In addition, the results obtained from the research paper can be used as a crucial reference in the development of WLED packages of higher quality.

REFERENCES


